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What are the physical properties of ammonia? Ammonia is a covalent

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compound with the following physical properties: Ammonia is very soluble in water, but it ionises partially in water to form a weak alkali. A  $0.1 \text{ mol dm}^{-3}$  ammonia solution has a pH of about 10.; Ammonia being alkaline can undergo neutralisation with acids to form ammonium salts.

Ammonia + Acid      Ammonium salt

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What are the physical properties of ammonia? - A Plus Topper

Solution: Calculating the amount of solute.

The number of moles of a solute in a given volume of solution can be calculated using the following equation. 1. Calculate the mass of hydrochloric acid in  $300 \text{ cm}^3$  of  $2.0 \text{ mol dm}^{-3}$  hydrochloric acid. [Relative atomic mass: H, 1; Cl, 35.5] Solution: 2.

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How to calculate concentration of ... - RS

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Aggarwal Solutions

Chlorine is the chemical name of Cl.

Chlorine is commonly used as antiseptic.

Visit BYJU'S to understand the properties, atomic mass of chlorine and uses of Chlorine (Cl) explained by the expert teachers.

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Chlorine (Cl) - Structure, Properties, Uses, and FAQs

Solution :  $a_n = a + (n - 1)d$   
 $0 = 27 + (n-1)(-3)$   
 $30 = 3n$   
 $n = 10$   
10th term. OR  $a_n = a + (n - 1)d$   
 $4 = a + 6(-4)$   
 $a = 28.$

Question 6. For what values of k, the equation  $9x^2 + 6kx + 4 = 0$  has equal roots?

Solution :  $9x^2 + 6kx + 4 = 0$   
 $(6k)^2 - 4 \times 9 \times 4 = 0$   
 $36k^2 = 144$   
 $k^2 = 4$   
 $k = \pm 2.$

Question 7. Find the roots of the equation  $x^2 + 1x \dots$

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When electricity is passed through such a solution, ions migrate towards oppositely charged electrodes and are discharged.  $2\text{H}^+ + 2\text{e}^- \rightarrow \text{H}_2(\text{g})$  At cathode  $2\text{Cl}^- \rightarrow \text{Cl}_2(\text{g}) + 2\text{e}^-$  At anode – Try Also : – Concise Selina Chemistry Solutions.  
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