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Stoichiometry: Iron is  
combined with

carbon in a series of  
reactions to form pig

iron, which is about

4.3% carbon.  $2C + O_2$

$2CO + Fe_2O_3 \rightarrow 3CO_2 + 2Fe$

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$3\text{CO}_2 + 2\text{CO} + \text{C}$  (in iron)

$\text{CO}_2$  Pig iron is easier to shape than pure iron, and the

presence of carbon lowers its melting point

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Form Goals To

explain why changes

that lead to greater

dispersal of matter

are likely to take

place. To explain why

liquids tend to mix

and form solutions.

To show how you can

predict whether

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soluble in water.

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Solids In your

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unit of an element

that still maintains

the properties that

element. ELEMENT—

a pure substance

made up of only one

type of atom.

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